

Bioorganic & Medicinal Chemistry Letters

Bioorganic & Medicinal Chemistry Letters 16 (2006) 3843-3846

Privileged structure based ligands for melanocortin receptors—4,4-Disubstituted piperidine derivatives

Steven L. Kuklish,* Ryan T. Backer, Karin Briner, Christopher W. Doecke, Saba Husain, Jeffrey T. Mullaney, Paul L. Ornstein, John M. Zgombick, Thomas P. O'Brien and Matthew J. Fisher

Lilly Research Laboratories, A Division of Eli Lilly and Company, Lilly Corporate Center, Indianapolis, IN 46258, USA Received 10 March 2006; revised 7 April 2006; accepted 7 April 2006 Available online 11 May 2006

Abstract—Homologation and cyclization back to the chiral methine of compound 3 yields achiral 4,4-disubstituted piperidine privileged structures (e.g., 8a) useful in the construction of melanocortin 4 receptor (MC4R) ligands. The piperidine nitrogen was replaced with carbon, oxygen, sulfur, and sulfone with minor erosion of binding. The methyl cyclohexane substituent was the most potent while significant affinity was still seen for smaller lipophilic groups such as ethyl. © 2006 Elsevier Ltd. All rights reserved.

The melanocortin 4 receptor (MC4R) is one of five type I G-protein-coupled receptors (GPCRs) which have the melanocortin peptides α -, β -, and γ -melanocyte-stimulating hormones (MSH) as their endogenous ligands. 1 The MC4R is expressed in the hypothalamus and is thought to regulate a variety of processes including feeding, metabolism, and reproductive behaviors.² We have been interested in developing ligands for the melanocortin receptors to facilitate their pharmacological characterization, and our efforts to date have yielded compounds primarily selective for MC4R. The compound series developed thus far have all required a dipeptide address element coupled to a C-terminal privileged structure cap (Fig. 1).³

Pharmacophore requirements for the piperazine-containing subunits have generally included both hydrophobic and polar moieties. We have found that a number of structurally related piperazine-containing moieties afford reasonable activity and a fair degree of interchangeability when attached to the dipeptide address element. Recently, we have reported on several new structurally related families, the latest of which is exemplified by structure 3.4,5 This scaffold uses a chiral

Keywords: Melanocortin; MC4; Privileged structure; G-protein coupled receptors; GPCR.

2 Dipeptide Address Element 1 MC₄ Ki = 5 nM 3 Priveleged Structure

Figure 1.

tertiary carbon atom to link the key lipophilic and polar pharmacophores.

While the activity achieved with these structures has been adequate for pharmacological characterization, the chiral center afforded an unnecessary level of complexity in synthesis. We wondered if this chiral carbon and attached polar moiety could be replaced with a symmetrical subunit that still allowed the desired privileged structure pharmacophores to be presented in a similar orientation. Specifically, we thought that cyclization of the polar residue back to the methine position of the chiral carbon of 3, affording an achiral 4,4-disubstituted piperidine (8a), might provide topography similar to our previously described privileged structures (Fig. 2).

^{*}Corresponding author. Tel.: +1 317 433 6519; e-mail: Kuklish_steven@lilly.com

Figure 2. The design strategy to mitigate asymmetry in our previous privileged structure was achieved by the cyclization and homologation, **a** and **b**, respectively.

Simple modeling of the 4,4-disubstituted piperidine of **8a** revealed that many of the conformations accessible to **3** remain available to the more constrained 4,4-disubstituted piperidine **8a**. Herein, we report on the reduction of this hypothesis to practice and the continued progress on the preparation and use of novel 4,4-disubstituted piperidine structures in the construction of ligands for the melanocortin receptors.

Preparations of structures 8a-l containing functionalized piperidines are outlined in Scheme 1. Intermediates 6a, b were conveniently prepared from the corresponding protected piperidines 4a, b using a modified Strecker synthesis.⁶ Treatment of amino nitrile **6a** with alkyl Grignards gave modest yields (10–50%) of the 4,4-disubstituted piperazines **7a–g**. The reactions of amino nitrile 6b with alkyl Grignards were more efficient for the preparation of intermediates 7h (86% yield) and 7i (94% yield) (Table 1). Deprotection of the Boc-protected piperazines 7a-g with TFA yielded the N-methyl piperidines 8a-g. Piperidine intermediates 7h (R = Boc) and 7i(R = Boc) were deprotected with TFA, then sulfonylated, acylated or reductively aminated to yield the corresponding benzyl-protected, functionalized 4,4-disubstituted piperidines. The functionalized piperidines were then debenzylated under palladium-mediated hydrogenolysis to yield structures 8h-l. Cyclic ketones **9a**–c were transformed in a similar manner as outlined for the above piperidine ketones (Scheme 2). Notably, reaction of the amino nitrile with an alkyl Grignard was again more efficient (91–98% in route to 10a–c) than observed for compounds 7a-g. Compounds that contain a basic piperidine nitrogen such as 7a-g were typically

Scheme 1. Reagents: (a) TMSCN, ZnI₂; (b) R²MgBr, THF; (c) TFA, DCM; (d) One of the following; MsCl, Alkyl-COCl, Alkyl-COH and NaBH₄; (e) Pd/C, H₂, EtOH.

Table 1.

Compound	P	R	\mathbb{R}^2	Compound
6a	Boc	Me	Ph	7a
6a	Boc	Me	Bn	7b
6a	Boc	Me	c-C ₆ H ₁₁ —CH ₂	7c
6a	Boc	Me	$n-C_6H_{13}$	7d
6a	Boc	Me	Et	7e
6a	Boc	Me	2-Propene	7 f
6a	Boc	Me	$(CH_3)_2CH$ — CH_2	7 g
6b	Bn	Boc	c-C ₆ H ₁₁ —CH ₂	7h
6b	Bn	Boc	$(CH_3)_2CH$ — CH_2	7i

Scheme 2. Reagents: (a) TMSCN, ZnI₂; (b) *i*Bu-MgBr, THF, 91–98%; (c) TFA, DCM; (d) MsOH, mCPBA, DCM.

less efficient in Grignard substitution reactions when compared to those without the basic amine. Oxidation of the thiomorpholine **10c** with mCPBA in the presence of methanesulfonic acid yielded **10d**. The piperazine structures **8a–1** and **10a–d** were coupled to the Boc-protected dipeptide^{9,10} **11** in the presence of HATU as previously described.¹¹ The penultimate compounds were then treated with TFA to generate **12a–p** as the corresponding TFA salts (Scheme 3).

Final products **12a**–**p** were evaluated for binding at the MC4 receptor by determining competitive inhibition of [125 I]NDP- α -MSH binding in HEK293 cells stably transfected with human MC4 receptors as previously described. 12,13 Compound **12a**, an achiral analog of **1**, yielded high affinity with a K_i of 7 nM. This observation suggests that this new achiral structure can mimic

Scheme 3. Reagents: (e) HATU, DIEA, DCM; (f) TFA, DCM.

the key topography afforded by chiral analog 1. Using 12a as a lead structure, we next explored the nature of the aliphatic group \mathbb{R}^2 , while holding the N-methyl piperidine constant (12a-g, Table 2). The aromatic counterpart to 12a, where $R^2 = Bn$ (12b), displayed a 6-fold decrease in binding affinity ($K_i = 44 \text{ nM}$). Removal of the methylene spacer of **12b** ($\mathbb{R}^2 = \mathbb{B}$ n) yields compound 12c, $(R^2 = Ph)$ whose K_i of 215 nM is 31-fold less than analog 12a. The alkyl substituted compound 12d $(R^2 = n\text{-hexyl})$ was 10-fold less active than compound **12a.** Contracting the aliphatic unit of **12d** ($R^2 = n$ -hexyl) to yield compound 12e (R^2 = ethyl) decreased binding affinity by 5-fold ($K_i = 370 \text{ nM}$). The isopropyl analog **12f** resulted in binding affinity 9-fold better ($K_i = 42 \text{ nM}$) than its non-branched counterpart 12e ($R^2 = ethyl$). Homologation of 12f afforded the isobutyl analog 12g $(R^2 = isobutyl)$ which had a 3-fold improvement in binding relative to 12f.

Next we turned our attention to functionalization of the piperidine nitrogen, while holding R^2 constant as isobutyl. We observed minimal effects on binding affinities for the family of alkyl substitutions on nitrogen. Both 12h (R = ethyl) and 12i (R = isopropyl) failed to differentiate from the parent compound 12g (R = methyl) with binding affinities of 16, 13, and 13 nM, respectively. Acylation of the nitrogen yielded 12j (R = acetyl) which resulted in a 3-fold loss in binding affinity ($K_i = 38$ nM) relative to 12g. The branched alkyl amide of 12k (R = COCH(CH₃)₂) demonstrated lower binding affinity ($K_i = 101$ nM) than its simple alkyl amine counterpart 12g.

The focus then turned to replacement of the piperidine ring with carbocycles and non-basic heterocycles (Table 3). Here we found that both all aliphatic carbocycle ($12m \ X = CH_2$) and heterocycles $12n-p \ (X = O, X = S, \text{ and } X = SO_2, \text{ respectively})$ produced binding affinities typically 2- to 6-fold less than their tertiary

Table 2.

Compound	R	\mathbb{R}^2	MC4 K _i (nM)
12a	Me	c-C ₆ H ₁₁ —CH ₂	6.8
12b	Me	Bn	43.9
12c	Me	Ph	214.6
12d	Me	n-C ₆ H ₁₃	69.6
12e	Me	Et	369.9
12f	Me	$(CH_3)_2CH$	41.8
12g	Me	$(CH_3)_2CH$ — CH_2	12.7
12h	Et	$(CH_3)_2CH$ — CH_2	15.6
12i	$(CH_3)_2CH$	$(CH_3)_2CH$ — CH_2	12.9
12j	$COCH_3$	$(CH_3)_2CH$ — CH_2	37.5
12k	$COCH(CH_3)_2$	$(CH_3)_2CH$ — CH_2	100.7
12l	SO ₂ CH ₃	$(CH_3)_2CH$ — CH_2	27.3

The K_i data are the average of at least two determinations with the average error of 15%.

Table 3.

Compound	X	MC4 K _i (nM)
12m	CH_2	46.9
12n	O	31.4
12o	S	68.4
12p	SO_2	75.8

The K_i data are the average of at least two determinations with an average error of 15%.

amine- and tertiary sulfonamide-containing counterparts 12g (R = methyl) and 12l (R = SO_2CH_3).

The above data demonstrate that replacement of the chiral structure 3 with a 4,4-disubstituted piperidine yields structures which can be employed in the construction of a new and physicochemically diverse set of MC4R ligands. Consistent with our hypothesis, the presumed decrease in conformational mobility afforded by the ring structure of 8a did not preclude this new scaffold from attaining the required conformation for binding to the MC4R. In addition to the elimination of a stereocenter, the synthetic route was straightforward with many of the piperazine fragments being prepared in three steps. For the MC4R, the amine moiety of the disubstituted piperidine can posses an alkyl group (12g, polar, charged) or be converted to a sulfonamide (121, polar, non-charged) with minimal loss of affinity. Similarly, this piperidine nitrogen can be replaced by carbon, oxygen, sulfide, and sulfone while still displaying good MC4R affinity.

In summary, we have described a convenient synthesis of useful 4,4-disubstituted piperidines and heterocycles as achiral privileged structures with widely varying physical properties for the preparation of MC4R ligands. Further, the subunits described here may afford an attractive path to probe numerous GPCRs given the highly conserved nature of the binding domains across GPCRs. Understanding of privileged structure topographical requirements and developing a collection of these subunits can present a significant opportunity to rapidly prepare ligands for the pharmacological exploration of a variety of GPCRs.

References and notes

- Cone, R. D.; Mounthoy, K. G.; Robbins, L. S.; Nadu, J. H.; Johnson, K. R.; Roselli-Rehfuss, L.; Mortund, M. T. Ann. N. Y. Acad. Sci 1993, 680, 342.
- (a) Wardlay, S. L. J. Clin. Endocrinol. Metab 2001, 86, 1442;
 (b) The Melanocortin Receptors; Cone, R. D., Ed.; Humana: Totowa, NJ, 2000.
- 3. For a discussion of privileged structures, see: (a) Evans, B. E.; Rittle, K. E.; Bock, M. G.; DiPardo, R. M.; Freiding-

- er, R. M.; Whitter, W. L.; Lundell, G. F.; Veber, D. F.; Anderson, P. S.; Chang, R. S. L.; Lotti, V. J.; Cerino, D. J.; Chen, T. B.; Kling, P. J.; Kunkel, K. A.; Springer, J. P.; Hirshfield, J. J. Med. Chem. 1988, 31, 2235; (b) Bondensgaard, K.; Ankersen, M.; Thogersen, H.; Hansen, B. S.; Wulff, B. S.; Bywater, R. P. J. Med. Chem. 2004, 47, 888; (c) Costantino, L.; Barlocco, D. Curr. Med. Chem. 2006, 13, 65.
- Briner, K.; Collado, I.; Fisher, F. J.; Garcia-Paredes, C.; Kuklish, S. L.; Mateo, M. I.; Ornstein, P. L.; De Frutos, O. J. Med. Chem. 2005, manuscript submitted for review.
- Fisher, M. J.; Backer, R. T.; Collado, I.; Frutos, O. D.; Husain, S.; Hsiung, H. M.; Kuklish, S. L.; Mateo, A.; Mullaney, J. T.; Ornstein, P. L.; Paredes, C. G.; O'Brian, T. P.; Richardson, T. I.; Shah, J.; Zgombick, J. M.; Briner, K. Bioorg. Med. Chem. Lett. 2005, 22, 4973.
- Evans, D. S.; Carroll, G. L.; Truesdale, L. K. J. Org. Chem. 1974, 38, 914.
- Palani, A.; Shapiro, S.; Clader, J. W.; Greenlee, W. J.; Cox, K.; Strizki, J.; Endres, M.; Baroudy, B. M. *J. Med. Chem.* 2001, 44, 3339.
- 8. General reaction of α-amino nitriles with alkyl Grignards: Preparation of 12e. To a solution of 4-(4-cyano-1-methyl-piperidin-4-yl)-piperazine-1-carboxylic acid *tert*-butyl ester (0.5 g, 1.6 mmol) in THF (4 mL) was added ethyl magnesium bromide 3 M in Et₂O (5.4 mL, 1.6 mmol) (additional alkyl Grignard may be added and heated to 40 °C to drive the reaction to completion if needed. This extra effort was most commonly used for examples with structures like 7a–g). The mixture was stirred for 18 h and then poured into water (50 mL). The mixture was then

- partitioned between brine and EtOAc. The organic phase was then washed with brine and concentrated to dryness. Purification by column chromatography (silica gel, 9:1 chloroform/methanol), Yield: 23.6% (0.12 g).
- For background on the exploration of address elements: Shi, Q.; Arnold, M. B.; Backer, R. T.; Briner, K.; Buckmaster, J. L.; Canada, E. J.; Doecke, C. W.; Fisher, M. J.; Hertel, L. W.; Honigschmidt, N.; Hsiung, H. M.; Husain, S.; Kuklish, S. L.; Martinelli, M. J.; Mullaney, J. T.; Ornstein, P. L.; Reinhard, M. R.; Richardson, T. I.; Rothhaar, R.; Shah, J.; Wu, Z.; Xie, C.; Zgombick, J. M. Bioorg. Med. Chem. Lett. 2006, 1, 2341.
- For the preparation of compound 11: Backer, R. T.; Briner, K.; Collado Cano, I.; De Frutos-Garica, O.; Doecke, C. W.; Fisher, M. J.; Garcia-Paredes, C.; Kuklish, S. L.; Mancoso, V.; Martinelli, M. J.; Mateo Herranz A. I.; Mullaney, J. T. Patent WO 02/059095 2002.
- Speicher, A.; Klaus, T.; Eicher, T. J. Prakt. Chem. I Chem-Ztg 1998, 340, 581.
- Richardson, T. I.; Ornstein, P. L.; Briner, K.; Fisher, M. J.; Backer, R. T.; Biggers, C. K.; Clay, M. P.; Emmerson, P. J.; Hertel, L. W.; Hsiung, H. M.; Husain, S.; Kahl, S. D.; Lee, J. A.; Lindstrom, T. D.; Martinelli, M. J.; Mayer, J. P.; Mullaney, J. T.; O'Brien, T. P.; Pawlak, J. M.; Revell, K. D.; Shah, J.; Zgombick, J. M.; Herr, R. J.; Melekhov, A.; Sampson, P. B.; King, C.-H. R. J. Med. Chem. 2004, 47, 744.
- 13. Data given are the average of at least two determinations with an average error of 15%. Biological testing was conducted on either the resulting TFA salt or the hydrochloride salt.